

№4

1. $W(O) = ?$

$$M_r(HClO) = 1 + 35 + 16 = 52$$

$$W(O) = \frac{16}{52} \cdot 100\% = 31\%$$

$$W(H) = \frac{1}{52} \cdot 100\% = 2\%$$

$$W(Cl) = \frac{35}{52} \cdot 100\% = 67\%$$

$$31 + 2 + 67 = 100\%$$

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4. 504

